SHORT COMMUNICATIONS

Reaction of Acetonitrile with Cyclic Boronic Esters Derived from *cis*-Acenaphthenediol

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Reactions of six-membered cyclic boronic esters with acetonitrile are known to give the corresponding 5,6-dihydro-4H-1,3-oxazines [1–3]. In the present communication we report on a similar transformation of five-membered cyclic boronic esters, 2-alkylace-naphtho[1,2-d][1,3,2]dioxaborolanes **I** and **II**, into 2-methyldihydrooxazole derivative **III**. The reaction is accompanied by partial hydrolysis of **III** to 1-amino-2-hydroxyacenaphthene (**IV**).

The composition of a mixture of products III and IV (TLC) and the structure of the components were confirmed by the IR and mass spectra. The mass spectrum contained the molecular ion peaks of compounds III and IV with m/z values of 209 and 185 and relative intensities of 14% and 32%, respectively. Also, an ion peak with m/z 168 (100%) was observed, which corresponds loss of acetonitrile molecule from the molecular ion of III or of NH₃ molecule from the molecular ion of IV. In the IR spectrum of mixture III/IV strong bands at 3300 (OH, NH), 1660 (C=N), and 1600 cm $^{-1}$ (C=C_{arom}) were present. Treatment of the product mixture (0.11 g) with aqueous potassium hydroxide gave amino alcohol IV (0.1 g) which was isolated as a dark brown powder, decomposing above 180°C. This reaction supports the general character and mechanism of transformation of cyclic boronic esters into N-heterocycles [3] and provides a new route to 1,2-aminoalcohols.

Initial esters **I** and **II** were synthesized by reaction of the corresponding acyclic boronic esters [4] with *cis*-1,2-dihydroxyacenaphthene [5] according to the general procedure described in [6].

2-Isopropyl-3a,9b-dihydroacenaphtho[1,2-d]-[1,3,2]dioxaborole (I). Yield 60%, oily liquid. ¹H NMR spectrum, δ , ppm: 0.87 d [6H, (CH₃)₂, J = 6.8 Hz], 1.79 m (1H, CHB), 6.07 s (2H, CHO), 7.61 m (6H, H_{arom}). Mass spectrum, m/z ($I_{\rm rel}$, %): 238 (35) M^+ , 152 (100) $[M-C_3H_7BO_2]^+$.

2-Isobutyl-3a,9b-dihydroacenaphtho[1,2-d]-[1,3,2]dioxaborole (II). Yield 60%, mp 67–68°C (from benzene). ¹H NMR spectrum, δ , ppm: 0.80 d (2H, BCH₂, J = 7.2 Hz), 0.86 d [6H, (CH₃)₂, J = 6.6 Hz], 1.83 m (1H), 6.10 s (2H, CHO), 7.61 m (6H, H_{arom}). Mass spectrum, m/z (*Irel*, %): 252 (44) M^+ , 152 (100) $[M-C_4H_9BO_2]^+$.

Reactions of cyclic boronic esters I and II with acetonitrile. Concentrated sulfuric acid, $10.6\,$ ml (0.2 mol), was slowly added in a dropwise manner to a solution of 0.01 mol of compound I or II in 150 ml of acetonitrile, and the mixture was refluxed for 4 h (on a water bath). Excess acetonitrile was distilled off on a rotary evaporator, and the viscous residue was diluted with $100\,$ ml of water and extracted with chloroform ($2\times50\,$ ml). The aqueous phase was treated with solid sodium hydroxide to pH 9–10 on cooling

I, $R = iso-C_3H_7$; II, $R = iso-C_4H_9$.

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with ice, the precipitate was filtered off, and the filtrate was extracted with chloroform $(4 \times 50 \text{ ml})$; the solvent was removed from the extract, and the residue was combined with the precipitate. We thus isolated 0.5 g of a mixture of compounds **III** and **IV** to which 50 ml of 25% aqueous potassium hydroxide was added, and the mixture was refluxed for 3 h. It was extracted with chloroform $(2 \times 25 \text{ ml})$, and the extract was evaporated to obtain amino alcohol **IV**.

The mass spectra (70 eV) were obtained on an MKh-1321 instrument. The ¹H NMR spectra were recorded on a Bruker AM-250 spectrometer from 10% solutions in CDCl₃ containing TMS as internal reference. The IR spectra were measured on a Specord 75IR instrument in mineral oil. TLC analysis was performed on Silufol UV-254 plates (eluent benzene-chloroform, 1:1; development with iodine vapor).

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